# organic papers

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# Da-Quan Peng,‡ Yun Liu, Zhi-Feng Lu and Jian-Hua Xu\*

Department of Chemistry, Nanjing University, Nanjing 210093, People's Republic of China

‡ Current address: Department of Chemistry, Chongqing Normal University, Chongqing 400047, People's Republic of China

Correspondence e-mail: xujh@nju.edu.cn

#### **Key indicators**

Single-crystal X-ray study T = 288 KMean  $\sigma$ (C–C) = 0.005 Å R factor = 0.053 wR factor = 0.152 Data-to-parameter ratio = 15.4

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# 2-Methyl-2-(trimethylsilyloxy)-3,4-dihydro-2*H*,5*H*-pyrano[3,2-c][1]benzopyran-5-one

In the title compound,  $\rm C_{16}H_{20}O_4Si,$  the coumarin moiety is essentially planar and the pyran ring adopts a half-chair conformation.

# Comment

*o*-Quinone methides constitute a class of highly reactive intermediates and their cycloaddition chemistry has attracted considerable interest (Nair *et al.*, 2001). In our studies of the cycloaddition reactions of *o*-quinone methides derived from 4hydroxycoumarin, we have found that the cycloaddition reaction between coumarin quinone methide and an excess amount of trimethyl[(1-methylethenyl)oxy]silane afforded the title compound, (I), as one of the products. As part of this study, we have undertaken the X-ray crystallographic analysis of (I) in order to elucidate the conformation and configuration of this cycloadduct product.



The bond lengths and angles in (I) (Fig. 1 and Table 1) are in good agreement with those found in related compounds (Savell *et al.*, 1989). The coumarin moiety is essentially planar, with atom O3 deviating from the mean plane by 0.152 (4) Å. The dihedral angle between the heterocyclic plane (C8/C9/O4/



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## Figure 1

The structure of (I), showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

Received 28 September 2005 Accepted 10 October 2005 Online 15 October 2005 C12/C11/C10) and the fused benzene ring is  $3.2 (3)^{\circ}$ . The pyran ring (C4/C6–C8/C10/O2) adopts a half-chair conformation, with atoms C4 and C6 deviating from the C7/C8/C10/O2 plane by 0.237 (4) and -0.487 (4) Å, respectively.

In the crystal structure, the molecular packing is stabilized by intermolecular  $C-H\cdots O$  interactions, as detailed in Table 2.

## **Experimental**

Under argon, a mixture of 4-hydroxycoumarin, 3 equivalents of trimethyl[(1-methylethenyl)oxy]silane and an excess amount of paraformaldehyde was suspended in dry dioxane and refluxed for about 14 h. Compound (I) was isolated by column chromatography of the reaction mixture on silica gel after evaporation of the solvent, in 76% yield. Single crystals of (I) were obtained by slow evaporation of a petroleum ether–ethyl acetate  $(3:1 \nu/\nu)$  solution of (I).

#### Crystal data

 $\begin{array}{l} C_{16}H_{20}O_4\text{Si} \\ M_r = 304.41 \\ \text{Monoclinic, } P2_1/c \\ a = 11.960 \ (2) \ \text{\AA} \\ b = 10.456 \ (2) \ \text{\AA} \\ c = 13.039 \ (3) \ \text{\AA} \\ \beta = 105.09 \ (3)^{\circ} \\ V = 1574.4 \ (6) \ \text{\AA}^3 \\ Z = 4 \end{array}$ 

### Data collection

Enraf–Nonius CAD-4 diffractometer  $\omega/2\theta$  scans Absorption correction:  $\psi$  scan (XCAD4; Harms & Wocadlo, 1995)  $T_{min} = 0.922$ ,  $T_{max} = 0.956$ 3075 measured reflections 2930 independent reflections

#### Refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.053$   $wR(F^2) = 0.152$  S = 1.002930 reflections 190 parameters H-atom parameters constrained  $D_x = 1.284 \text{ Mg m}^{-3}$ Mo  $K\alpha$  radiation Cell parameters from 25 reflections  $\theta = 10-13^{\circ}$   $\mu = 0.16 \text{ mm}^{-1}$  T = 288 (2) KBlock, colourless  $0.40 \times 0.31 \times 0.28 \text{ mm}$ 1878 reflections with  $I > 2\sigma(I)$   $R_{\text{int}} = 0.028$   $\theta_{\text{max}} = 25.5^{\circ}$   $h = 0 \rightarrow 14$   $k = 0 \rightarrow 12$  $l = -15 \rightarrow 15$ 

3 standard reflections every 200 reflections intensity decay: none

$w = 1/[\sigma^2(F_o^2) + (0.055P)^2]$
+ 1.48P]
where $P = (F_0^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{\rm max} < 0.001$
$\Delta \rho_{\rm max} = 0.29 \text{ e } \text{\AA}^{-3}$
$\Delta \rho_{\rm min} = -0.25 \text{ e} \text{ Å}^{-3}$

#### Table 1

Selected geometric parameters (Å, °).

01-C4 02-C4 03-C9	1.391 (4) 1.464 (4) 1.215 (4)	C4-C6 C6-C7	1.507 (5) 1.519 (5)
C4-O1-Si1	134.8 (2)	O1-C4-C5	113.0 (3)

# Table 2

H	lyd	lrogen-	bond	geometry	(A,	°).
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$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$\begin{array}{c} \text{C5-H5} C \cdots \text{O3}^{\text{i}} \\ \text{C13-H13} \cdots \text{O3}^{\text{ii}} \end{array}$	0.96 0.93	2.47 2.56	3.426 (5) 3.492 (4)	171 175
Symmetry codes: (i) x.	$-v + \frac{1}{2}, z + \frac{1}{2};$	(ii) $-x + 3, y - $	$\frac{1}{2}, -z + \frac{3}{2}$	

H atoms were included in the riding-model approximation, with C-H distances of 0.93, 0.96 and 0.97 Å for aromatic, methyl and methylene H atoms, respectively, and with  $U_{\rm iso}({\rm H}) = 1.2 U_{\rm eq}({\rm aromatic})$  and methylene C) or  $1.5 U_{\rm eq}({\rm methyl} {\rm C})$ .

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2003).

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